Stability indicating RP-LC-PDA method for the quantitative analysis of saxagliptin in pharmaceutical dosage form

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Saxagliptin is a potent and selective inhibitor of the enzyme dipeptidyl peptidase 4. It is effective in the treatment of type 2 diabetes mellitus because it stimulates the pancreas to produce insulin. In the present study, a liquid chromatography method was developed and validated to quantify the drug in tablets. This method was based on the isocratic elution of saxagliptin, using a mobile phase consisting of 0.1% phosphoric acid at pH 3.0 – methanol (70: 30, v/v) at a flow rate of 1 mL.min⁻¹ with UV detection at 225 nm. The chromatographic separation was achieved in 8 minutes on a Waters XBridge C18 column (250 mm x 4.6 mm, 5µm) maintained at ambient temperature. The proposed method proved to be specific and robust for the quality control of saxagliptin in pharmaceutical dosage forms, showing good linearity in the range of 15.0 – 100.0 µg.mL⁻¹ (r>0.999), precision (RSD<1.49%) and accuracy values between 99.42 and 101.59%. The method was found to be stability indicating and was successfully applied for the analysis of saxagliptin in tablets in a routine quality control laboratory.


INTRODUCTION

Saxagliptin (SAX, BMS-477118; Figure 1), is a potent, selective, long-acting and reversible inhibitor of the enzyme dipeptidyl peptidase 4 (DPP 4) used for the treatment of type 2 diabetes mellitus. Inhibition of this enzyme results in increased concentrations of the glucagon-like peptide 1 (GPL-1) and other endogenous incretins and consequently increases the stimulation of insulin release from the pancreas (Augeri et al., 2005).

Saxagliptin belongs to a group of anti-diabetic drugs, the DPP 4 inhibitors, along with sitagliptin and vildagliptin, but the former appears to be more potent since it is effective in the 2.5-10.0 mg range once a

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day compared with the therapeutic dose of 100 mg of sitagliptin (Deacon; Holst, 2009). Furthermore, saxagliptin can be used as monotherapy or combined with metformin, glyburide or thiazolidinedione (Karyekar et al., 2011).

In 2009, saxagliptin was approved by the European Medicines Agency (EMA) for the treatment of type 2 diabetes mellitus (EMA, 2009, 2013). Since then, bioanalytical HPLC methods have been reported to quantify saxagliptin and its major pharmacologically active metabolite, 5-hydroxy saxagliptin, in human plasma, which were applied to pharmacokinetic studies (Patel et al., 2010; Xu et al., 2012). Gao et al. (2012) developed and validated a method by UPLC-MS/MS for quantification of this pharmaceutical substance in plasma. The applicability of the method was demonstrated in a pharmacokinetic study in healthy rats. Nevertheless, the previously reported methods do not deal with the determination of the drug in pharmaceutical dosage form and are not suitable for quality control routine analysis. For pharmaceutical formulation, Kalaichelvi and Jayachandran (2011) developed a UV spectrophotometric method to determine saxagliptin in tablets. However, due to limitation of specificity, it could not be applied to the analysis of stability samples in the pharmaceutical industry, where a stability indicating method is required. Moreover, no official method has been described in the literature for the determination of saxagliptin in pharmaceutical dosage form by liquid chromatography. Therefore, the purpose of this study was to develop and validate a stability indicating reversed-phase liquid chromatography method with UV detection for the quantitative analysis of saxagliptin in pharmaceutical formulation.

MATERIAL AND METHODS

Chemicals

Saxagliptin reference substance was obtained from Ontario Chemicals (Ontario, Canada) and was certified to be 98.0 % pure. Tablets containing 5 mg of saxagliptin (Onglyza™, Bristol-Myers Squibb, New York, NY, USA) were obtained from commercial sources. Methanol (HPLC grade) was obtained from Tedia (Fairfield, USA) and phosphoric acid from Vetec (Rio de Janeiro, Brazil). Ultrapure water was obtained from Mega Purity System (Canada) water purification unit.

Instrumentation and analytical conditions

A Shimadzu LC system (Shimadzu, Kyoto, Japan) was used equipped with an LC-20AT pump, a SPD-M20A photodiode array (PDA) detector, a CBM-20A system controller, a DGU-20A5 degasser and a Rheodine® (Rohnert Park, CA, USA) injector model 7725i. The detector was set at 225 nm and peak areas were integrated automatically by computer using an LC Solution software program (version 1.24 SP1).

The chromatographic system used a Waters XBridge C18 (250 mm x 4.6 mm, 5 µm) column at ambient temperature. Separation was achieved using a mobile phase consisting of 0.1 % phosphoric acid at pH 3.0-methanol (70:30, v/v) at a flow rate of 1.0 mL.min$^{-1}$ within a runtime of 10 minutes. The injection volume was 20 µL.

Saxagliptin reference solution

The reference stock solution of saxagliptin was prepared by dissolving an accurately weighed 10 mg of reference substance in 20 mL of methanol acidified medium with 0.4 mL of 0.1 % phosphoric acid at pH 3.0. This stock solution was stored at 2-8 ºC. Further dilutions were made with water until the final concentration of 50.0 µg.mL$^{-1}$ for injection.

Preparation of tablet sample solution

For the preparation of the sample solution, tablets containing 5 mg of saxagliptin were accurately weighed and crushed. An appropriate amount of the tablet powder (equivalent to 2.5 mg of saxagliptin) was weighed and transferred into a 25 mL volumetric flask with 15 mL of methanol and 0.1 mL of 0.1 % phosphoric acid at pH 3.0. The sample was sonicated for 10 minutes and the flask volume was completed with methanol. Other dilutions were made after centrifugation, with water to reach a test concentration of 50.0 µg.mL$^{-1}$ saxagliptin. The solutions were then filtered through a 0.45 µm membrane filter (Sartorius Stedim Biotech, Germany).

Validation of the method

The method was validated using tablet samples of
saxagliptin by determining the following parameters: specificity, linearity, precision, accuracy and robustness, according to ICH guidelines (ICH, 2005).

Specificity

Forced degradation studies were performed by dissolving 10 mg of saxagliptin in 50 mL of methanol and later 5 mL of this sample (equivalent to 1000 µg of saxagliptin) was separately diluted with 5 mL of 1 mol.L⁻¹ hydrochloric acid and 5 mL of 0.1 mol.L⁻¹ sodium hydroxide. The studies under acid and alkaline conditions were carried out at 60 ºC for 2 hours and at ambient temperature for 10 minutes, respectively, neutralized and then diluted with water in a 20 mL volumetric flask to achieve a concentration of 50.0 µg.mL⁻¹. The oxidative condition was obtained by contacting the sample diluted in methanol at 1000 µg.mL⁻¹ saxagliptin with 5 mL of 30 % hydrogen peroxide, at ambient temperature. After 24 hours, protected from light, the sample was diluted with water to a concentration of 50.0 µg.mL⁻¹. Finally, for photodegradation, the amount of the tablet powder was weighed and transferred into a 10 mL volumetric flask with methanol, from which 1 mL was removed and exposed in a photostability chamber to 254 nm for 3 hours and 30 minutes. After exposure, the volume was transferred again to a 10 mL volumetric flask for concentration 50.0 µg.mL⁻¹.

Linearity

Linearity was determined by constructing three analytical curves, each one with five reference concentrations of saxagliptin in the 15.0-100.0 µg.mL⁻¹ range prepared by diluting an appropriate quantity of reference stock solution with water. Each concentration was injected in triplicate.

Precision

The precision of the analytical method is evaluated by repeatability (intra-day precision) and intermediate precision (inter-day precision) in the sample solutions. Repeatability was calculated by assaying six samples with a concentration of 50.0 µg.mL⁻¹, prepared according to the preparation of the sample solution, all on the same day and under the same experimental conditions. The intermediate precision was achieved from results obtained on three different days (six samples on the first day and three samples on the second and third days) and by another analyst in the same laboratory. The test was determined by calculating relative standard deviation (RSD).

Accuracy

Accuracy was assessed by determining the ability of the method to recover three different concentrations (corresponding to 80, 100 and 120% of the nominal analytical concentration) by the addition of known amounts (20.0, 30.0 and 40.0 µg.mL⁻¹) of the reference solution to the sample solution.

Robustness

The robustness of the method was evaluated by assaying the test solutions after slight changes in the analytical method. The factors chosen for this study were the composition of the mobile phase (phosphoric acid - methanol, 75 : 25 and 65 : 35, v/v), pH of phosphoric acid (2.5 and 3.5) and flow rate (0.8 and 1.2 mL.min⁻¹). The objective was to analyse any impact of each changed condition on method reliability for the routine quality control analysis.

RESULTS AND DISCUSSION

During the process of method development, several trials were performed testing different columns, different pH values and different proportion of the mobile phase. A good peak shape was observed when a Waters XBridge C18 (250 mm x 4.6 mm, 5 µm) column was used, which showed efficiency. A mixture of 0.1% phosphoric acid at pH 3.0-methanol (70:30, v/v) was adopted because saxagliptin is very soluble at low pH (EMA, 2009, 2013). These chromatographic conditions were optimized to separate any degradation products from the saxagliptin peak.

Method validation

Specificity

The forced degradation was performed to show that the developed method is able to separate the active from possible degradation products. Degradation was performed under basic, acid, oxidative and light exposure. When submitted to basic condition, the chromatogram showed three possible degradation products close to 3.5 minutes, 5 minutes and 6.5 minutes, while the saxagliptin content decreased about 11.1%. Under acid condition, was found 13.52% reduction in peak area and no additional peaks were detected. The oxidative condition produced two new peaks close to 5 minutes and the saxagliptin content decreased about 38%. When the samples were exposed to UVC radiation, no degradation peaks were detected and the recovery content was around 61%
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(Figure 2). The placebo was prepared and injected in order to compare it with the tablet sample (Figure 3). The studies with the PDA detector showed that the degradation products produced as a result of stress did not interfere in the detection of saxagliptin, with purity peak values greater than 0.9999 and demonstrating the ability to separate the drug from their degradation products. Therefore, it may be employed for analysis of stability samples of saxagliptin, even though the characterization of degradation products was not carried out.

**Linearity**

Five reference solutions were prepared containing 15.0, 25.0, 50.0, 75.0 and 100.0 µg.mL⁻¹ saxagliptin, corresponding to 30, 50, 100, 150 and 200% of the test concentration, respectively. Three curves were constructed by plotting the solution concentration against the peak area. Results are shown in Table I. The validity of the assay was verified by analysis of variance, which showed that there were linear regression and no deviation from linearity ($p < 0.05$).

**Precision**

Precision was estimated by repeatability and intermediate precision. The repeatability was assessed by analysing a solution of 50.0 µg.mL⁻¹ saxagliptin ($n = 6$) on the same day and under the same conditions. The intermediate precision ($n = 12$) was obtained by another analyst in different day. The RSD values are shown in Table II.

**Accuracy**

The accuracy of the RP-LC method was achieved by calculating percentage recovery of the added saxagliptin reference solution. The mean accuracy was 100.72% with an RSD of 0.37%. These values demonstrated that the method is accurate within the desired range (Table III).
Robustness
In all deliberately varied chromatographic conditions, the determination of the drug in the pharmaceutical formulation was not significantly altered and only resulted in a change in the retention time. The values obtained for the theoretical plates and the asymmetry were also satisfactory. The results are demonstrated in Table IV.

### TABLE II - Method repeatability and intermediate precision for saxagliptin

<table>
<thead>
<tr>
<th>Level (%)</th>
<th>Intra-day precision</th>
<th></th>
<th>Inter-day precision</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Mean (%)</td>
<td>RSD (%)</td>
<td>Another analyst; Mean of three days</td>
</tr>
<tr>
<td>Day 1 ((n = 6))</td>
<td>101.98</td>
<td>1.49</td>
<td></td>
</tr>
<tr>
<td>Day 2 ((n = 3))</td>
<td>99.56</td>
<td>1.20</td>
<td></td>
</tr>
<tr>
<td>Day 3 ((n = 3))</td>
<td>100.84</td>
<td>1.35</td>
<td></td>
</tr>
<tr>
<td>Inter-day precision</td>
<td>100.79</td>
<td>1.20</td>
<td></td>
</tr>
</tbody>
</table>

### TABLE III - Results of accuracy by HPLC method

<table>
<thead>
<tr>
<th>Level (%)</th>
<th>Amount of the drug ((\mu g.mL^{-1}))</th>
<th>Recovery (%)</th>
<th>RSD (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>added</td>
<td>found</td>
<td></td>
<td></td>
</tr>
<tr>
<td>80</td>
<td>15.28</td>
<td>15.48</td>
<td>101.33</td>
</tr>
<tr>
<td>19.80</td>
<td>20.08</td>
<td>101.42</td>
<td>0.32</td>
</tr>
<tr>
<td>19.50</td>
<td>19.63</td>
<td>100.70</td>
<td></td>
</tr>
<tr>
<td>100</td>
<td>25.47</td>
<td>25.81</td>
<td>101.37</td>
</tr>
<tr>
<td>30.27</td>
<td>30.33</td>
<td>100.95</td>
<td>0.32</td>
</tr>
<tr>
<td>28.51</td>
<td>40.25</td>
<td>101.59</td>
<td></td>
</tr>
<tr>
<td>120</td>
<td>35.65</td>
<td>35.63</td>
<td>99.56</td>
</tr>
<tr>
<td>38.28</td>
<td>38.05</td>
<td>99.42</td>
<td></td>
</tr>
</tbody>
</table>

### TABLE IV - Robustness results of saxagliptin by the developed method

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Value</th>
<th>Assay (%)</th>
<th>%RSD</th>
<th>RT(^a) (min)</th>
<th>Theoretical Plates</th>
<th>Asymmetry</th>
</tr>
</thead>
<tbody>
<tr>
<td>Flow rate ((mL.min^{-1}))</td>
<td>0.8</td>
<td>99.57</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>1.0</td>
<td>99.48</td>
<td>0.57</td>
<td>10.06</td>
<td>7693</td>
<td>1.09</td>
</tr>
<tr>
<td></td>
<td>1.2</td>
<td>100.50</td>
<td></td>
<td>8.12</td>
<td>7231</td>
<td>1.12</td>
</tr>
<tr>
<td></td>
<td>2.5</td>
<td>98.81</td>
<td></td>
<td>7.17</td>
<td>6316</td>
<td>1.12</td>
</tr>
<tr>
<td>0.1% phosphoric acid</td>
<td>3.0</td>
<td>98.36</td>
<td>0.31</td>
<td>8.08</td>
<td>6943</td>
<td>1.06</td>
</tr>
<tr>
<td></td>
<td>3.5</td>
<td>98.23</td>
<td></td>
<td>8.62</td>
<td>7245</td>
<td>1.06</td>
</tr>
<tr>
<td>Proportion of mobile phase</td>
<td>75 : 25</td>
<td>98.75</td>
<td></td>
<td>10.20</td>
<td>7130</td>
<td>1.07</td>
</tr>
<tr>
<td></td>
<td>70 : 30</td>
<td>100.56</td>
<td>1.11</td>
<td>8.28</td>
<td>7205</td>
<td>1.10</td>
</tr>
<tr>
<td></td>
<td>65 : 35</td>
<td>100.77</td>
<td></td>
<td>7.83</td>
<td>7168</td>
<td>1.18</td>
</tr>
</tbody>
</table>

\(^a\)RT: retention time
System suitability

The suitability of the chromatographic system was tested before starting the work to evaluate the asymmetry, number of theoretical plates and capacity factor ($k'$), with average results of 1.11, 7,805 and 2.25, respectively. The RSD of the peak area was 1.25%. The experimental results were within the acceptable values, indicating that the system is suitable for the analysis intended.

CONCLUSION

The developed method was found to be precise with RSD values for intra-day and inter-day precision studies less than 1.49%. Good recoveries for the drug were obtained at each added concentration, indicating that the method was accurate. During robustness check, the RSD and percentage of drug content is well within the acceptance criteria. Specificity was indicated by the resolution of saxagliptin peak from the peaks of degradation product. The peak purity profile by PDA detector confirmed the specificity without any interference from the degradation products. Hence, the proposed RP-LC method can be successfully applied during stability studies of saxagliptin in pharmaceutical dosage form.

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REFERENCES


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